

Noval Route through O–C Bond Formation for the Synthesis of Diastereomeric 2, 4-Disubstituted Pyrano [2, 3-B] Quinolines from 3-Formyl-2-Quinolones via Intramolecular Electrophilic Cyclization

Dr. Shraddha Upadhyay*

Author Affiliations:

Head, Department of Chemistry, Swami Vivekanand Subharti University, Meerut, Uttar Pradesh 250005, India.

***Corresponding Author:**

Dr Shraddha Upadhyay, Head, Department of Chemistry, Swami Vivekanand Subharti University, Meerut, Uttar Pradesh 250005, India

E-mail: dr.shrupa@gmail.com

Received on 27th January 2018

Accepted on 16th July 2018

Abstract

One-pot, One-step synthesis of 2, 4-disubstituted pyrano[2,3-b]quinolines from 3-formyl-2-quinolones is described from 3-formyl-2-quinolones via intramolecular electrophilic cyclization.

Keywords: Electrophilic cyclization, Pyranoquinoline, Cycloaddition reactions

Many alkaloids such as flindersine, oricine and verprisine [1] which are from Rutaceae family [2], having Pyranoquinoline moiety. They possess broad range of bioactivity such as anti-allergic, psychotropic, anti-inflammatory and estrogenic activities [3]. and have wide applications as drugs, pharmaceuticals and agrochemicals. They have great attention in synthetic as well as medicinal field. Thus, to discover new method for the synthesis of this compound attracts much attention. Although various methods for their synthesis and their annulated analogues described in the literature [4].

The most common method for the synthesis of tri- and tetracyclic pyranoquinolines [5] are cycloaddition reactions.

Bhuyan et al. for the synthesis of tetracyclic pyranoquinolines have reported an intramolecular 1, 3-dipolar cycloaddition reaction [6]. Using 1,3 dipoles such as nitrones, nitrile imines, nitrile oxides [5], (m.s paper), polyphosphoric acid [2e], DDQ [6], and Prevost reaction [7].

Reaction next to cycloaddition reactions is acid-catalyzed cyclization for the syntheses of pyranoquinoline derivatives.

But all these conditions possess drawbacks such as expensive starting materials and poor yields of both starting materials and final products. Thus, to overcome these difficulties it is essential to develop new and efficient synthetic routes for the preparation of this class of compounds.

Precursor, 2-chloro-3-formylquinolines 1, [8] is synthesized via a Vilsmeier–Haack approach using acetanilide, DMF and POCl₃. We have used 3-cyanoquinoline and 3-methoxyquinoline derivative of quinoline for the synthesis of annulated carbocycles [9], sulfur, and nitrogen-containing heterocycles [10]. All these possess broad range of bioactivities.

Dr. Shraddha Upadhyay / Noval Route through O–C Bond Formation for the Synthesis of Diastereomeric 2, 4-Disubstituted Pyrano [2, 3-B] Quinolines from 3-Formyl-2-Quinolones via Intramolecular Electrophilic Cyclization

Further, we have synthesis stereo controlled cis-disubstituted pyrano[2,3-b] in continuation of our research from 3-formyl-2-quinolones 2.

Precursor, 3-formyl-2-quinolones has been synthesis via intramolecular electrophilic cyclization through O–C bond formation from 3-formyl-2-quinolines 1, (Scheme 2, Table 1).

Table 1: Synthesis of diastereomeric cis/trans-4-hydroxy-2-iodomethylpyrano [2,3-b]quinolines from 3-homoallyl-2-quinolones 3

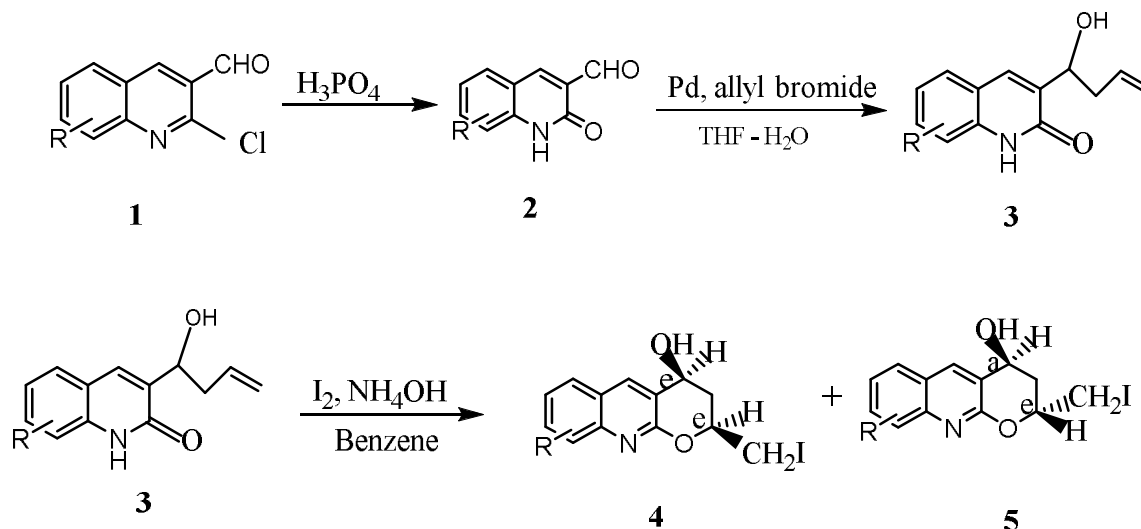
Entry	Substrate	R	Time (h)	Product	Ratio (cis/trans)	Yield (%)
1	3a	H	4	4a/5a	77:23	88
2	3b	6Me	2.5	4b/5b	90:10	81
3	3c	7Me	4	4c/5c	79:21	84
4	3d	6MeO	3	4d/5d	84:16	83
5	3e	7MeO	3.5	4e/5e	100	85
6	3f	8Et	2.5	4f/5f	100	88

Different analogous have been synthesis from 2-chloro-3-formylquinolines. via different routes[5,11].

3-Formyl-2-quinolones 2 were prepared by refluxing 2-chloro-3-formylquinolines 1 in aqueous acetic acid. Product is obtained in quantitative yield.

Further, allylation of 3-Formyl-2-quinolones using allyl indium bromide and DMF, give 3-homoallyl-2-quinolones 3 at room temperature in excellent yields (89–94%). The structure of compound 3a analyze from spectral analysis.

Compound 3a when treated with I₂ and sodium bicarbonate using THF as solvent at room temperature provide a mixture of 4a/5a, 14 in 88% yield (Scheme 1, 2). In mixture of compound cis for is more dominating 4a which is illustrated out through spectral analysis (¹H NMR spectroscopy).



Diastereoisomers 4a/5a is not separated via column chromatography.

In, spectral analysis C-5 proton (d = 8.32) of cis-4a due to hydroxy group occupies an equatorial position in the cis-conformation appear at downfield in NMR scale while that of trans-isomer 5a, appear at d = 8.12.

Same pattern of observation were also seen in the spectral data of C-2 and C-4 in the cis-isomers, which appeared downfield in comparison to the trans-iso isomers.

Non-equivalent proton of CH₂I in cis-compounds protons appear as quartet.

Further, molecular models shows that most stable conformation of the cis-isomers are that which have hydroxy group at C-4 and the iodomethyl group at C-2 in equatorial positions.

Cis and Trans isomers in **Table 1**, entries 1–4 is non separable.

In conclusion, via using inexpensive and readily available starting materials and reagents, we have synthesis 4-hydroxy-2-iodomethyl-3, 4-dihydro-2H-pyrano [2, 3-b]-quinolines in good yield. Procedure requires easy work-up and structure of for compound characterized by spectral analysis.

REFERENCES

1. (a) Yamada, N.; Kadowaki, S.; Takahashi, K.; Umezu, K. *Biochem. Pharmacol.* 1992, 44, 1211–1213; (b) Faber, K.; Stueckler, H.; Kappe, T.J. *Heterocycl. Chem.* 1984, 21, 1177–1181; (c) Johnson, J.V.; Rauckmann, B.S.; Baccanari, D.P.; Roth, B. *J. Med. Chem.* 1989, 32, 1942–1949; (d) Nesterova, I.; Alekseeva, L.M.; Andreeva, L.M.; Golovira, S.M.; Granic, V.G. *Khim. Farm. Zh.* 1995, 29, 31–34; (e) Rajendra Prasad, K.J.; Sekar, M.J. *Nat. Prod.* 1998, 61, 294–296
2. (a) Michael, J.P. *Nat. Prod. Rep.* 1997, 14, 605–618; (b) Balsubramanian, M.; Keay, J.G. In *Comprehensive Heterocyclic Chemistry II*; Katritzky, A. R., Rees, C. W., Scriven, E.F.V., Eds.; Pergamon Press: Oxford, 1996; Vol. 5, Chapter 5.06, p 245; (c) Wright, C. W.; AddacKyereme, J.; Breen, A. G.; Brown, J.E.; Cox, M.F.; Croft, S.L.; Gokcek, Y.; Kendrick, H.; Phillips, R.M.; Pollet, P.L. *J. Med. Chem.* 2001, 44, 3187–3194; (d) Sahu, N.S.; Pal, C.; Mandal, N.B.; Banerjee, S.; Raha, M.; Kundu, A.P.; Basu, A.; Ghosh, M.; Roy, K.; Bandyopadhyay, S. *Bioorg. Med. Chem.* 2002, 10, 1687–1693; (e) Kournetsov, V.V.; Mendez, L. Y.V.; Gomez, C. M.M. *Curr. Org. Chem.* 2005, 9, 141–161.
3. (a) Desai, M.C.; Thadeio, P. F. *Tetrahedron Lett.* 1989, 30, 5223–5226; (b) Ashrof, M. A.; Raman, P. S. *J. Indian Chem. Soc.* 1992, 69, 690–691; (c) Cziaky, Z.; Sebok, P. J. *Heterocycl. Chem.* 1994, 31, 701–705.
4. (a) Mahesh, C.J.; Makesh, S.V.; Perumal, P.T. *Bioorg. Med. Chem. Lett.* 2004, 14, 2035–2040; (b) Yadav, J. S.; Reddy, B.V.S.; Rao, R.S.; Kumar, S.K.; Kunwar, A C. *Tetrahedron* 2002, 58, 7891–7896; (c) Mahesh, M.; Reddy, Ch. V.; Reddy, K.S.; Raju, P.V.K.; Reddy, V.V.N. *Synth. Commun.* 2004, 34, 4089–4104.
5. Bhuyan, P.J.; Baruah, B.; Kalita, P.K. *Tetrahedron Lett.* 2006, 47, 7779–7782.
6. Plozzi, F.; Venturella, P.; Bellino, A. *Gazz. Chim. Ital.* 1969, 99, 711–714.
7. Subramanian, M.; Mohan, P. S.; Shanmugam, P.; Rajendra Prasad, K. *J. Z. Naturforsch.* 1992, 47b, 1016–1020.
8. Singh, R.M.; Srivastava, A. *Indian J. Chem.* 2005, 44B, 1868–1875.
9. Singh, R.M.; Chandra, A.; Singh, M.K. *Synth. Commun.* 2007, 37, 1689–1695.
10. (a) Singh, R.M.; Chandra, A.; Srivastava, A. *Indian J. Chem.* 2005, 44B, 2077–2084; (b) Singh, R. M.; Singh, M.K.; Srivastava, A. *Indian J. Chem.* 2006, 45B, 292–296; (c) Singh, R. M.; Chandra, A.; Srivastava, A. *Indian J. Chem.* 2007, 46B, 303–307.
11. Szabo, Z.; Cziaky, Z.J. *Heterocycl. Chem.* 1995, 32, 755–760.