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Effect Of An Inorganic Compound On Geopolymer Concrete And Ordinary Portland Cement Concrete

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Abstract

The main problems with durable concrete constructions are attacks by sulphates and chlorides on concrete. Thus, the impact of concrete strength characteristics is based on the addition of fly ash exposed to the environment used as sulphate and chloride solution as the primary goal of this research work. In this research work, geopolymer concrete (GPC) is used instead of ordinary Portland cement (OPC). The concrete composition is prepared with geopolymer varies. Samples are demolded and then submerged in water for a full 28 days to cure. Following this, the samples are placed in different solutions of 10% sodium chloride (NaCl) and 10% sodium sulphate (Na₂SO₄) for hardening times at 28, 56, and 90 days respectively. A degree of damage, variation in compressive strength, and weight change were used to assess the impacts of sulphate and chloride on the concretes. After 90 days, the exposure of fly ash to concrete significantly improved its compressive strength at 33.11% as compared to the strength of 10% NaCl solution, water and OPC, respectively. In contrast to the OPC, the 10% NaCl solution decreased the compressive strength of fly ash-containing GPC after 90 days of exposure. In comparison to other models, the maximum ultrasonic pulse velocity of GPC-7 was attained at 4430 m/s. The minimal charge cleared in the GPC-7 fast chloride permeability test is 1045 coulombs. According to this study, adding fly ash to concrete as an additional cementitious material may help lessen the harmful impacts of sulphate and chloride salts. The study's findings suggest that adding ground fly ash to concrete as an additional cementitious element strengthens the material's durability to harsh environments.

Keywords: Geopolymer Concrete; Curing; Compressive Strength; Sodium sulphate and chloride; Ordinary Portland cement.

1. Introduction

The formation of concrete composite materials is consisting of cement, sand as a fine aggregate, crushed stone as a coarse aggregate, and water. It may be readily worked into a variety of shapes because to its workability. Concrete also offers excellent fire resistance and compressive strength [1, 2]. The concrete is classified into two categories, the lighter, less dense type weighs less than 19,200 kg/m³, and the more typical dense type weighs 24,000 kg/m³. The cement is a major component of concrete. Tricalcium silicate (3CaOS1O₂), dicalcium silicate (2CaCOS1O₂), tricalcium aluminate (3CaOAlO₃), and tetra calcium alumino ferrite (4CaOAl₂O₃ Fe₂O₃) are main chemical components of Portland cement. Concrete materials are subjected to elements known as contaminants or impurities, which can impact the plain and reinforced concrete strength. The environmental pollutants are acids,

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chlorides, sulphates, organic debris, etc. [3]. This study focuses on the highly concerning effect of chloride. When mixed with the ingredients of concrete, sodium chloride (NaCl) can cause harm, according to Helmenstine [4] and Robertson [5], but it can also speed up the pigments in the wet concrete's surface right away. Carter [6] found in his studies conducted between 1993 and 2005 that sodium chloride corrodes concrete more than other substances, but Lee et al.. [7] assert that the solution as NaCl affect the concrete.

Concrete may suffer negative effects from industrial effluents' chemical contaminants. This raises serious questions about the characteristics of finished concrete and the mixing of water during cement hydration. Two sources of water are often used in the making of concrete: the moisture in the particles and the water that is added. The majority of water quality criteria focus more on the amount of water added than the fine and coarse aggregate moisture content. Most of the time, people overlook the quality of the water because they assume that adding drinkable water will neutralize any contaminants. Geographically varying quantities of dissolved inorganic compounds are predicted in groundwater, which mostly originates from minerals. Salts of calcium, magnesium, potassium, and other elements may be found in water.

Concrete's resistance to chloride attack is compromised. Previous studies [8-13] that looked at how concrete affected reinforced concrete concluded that precautions against chloride attack are especially crucial because corrosion is the major reason for reinforcement, which accounts are more than 40% of structural failures.

However, the current investigation focuses on how mass concrete is affected by water contaminated with chloride. The goals are to ascertain the strengths of concrete produced using water containing varied percentages of chloride contamination, to investigate the structural failure pattern that may result from this contamination, and to formulate results. In this research study focused on mechanical qualities, specifically compressive strength, and the ability of geopolymer concrete to withstand sulphate assault.

2. Related Works

Research is required to determine how chemically contaminated water affects the strength and longevity of hardened concrete. Results of a study on the effects of different deicing agents and exposure times on concrete materials showed that different deicing agents saw varying rates of penetration into a particular paste, which led to varying degrees of damage to the concrete [13]. provided an evaluation of the chemical resistance of eight distinct polymeric mortar formulations [14]. Reported on how environmental variables affected the durability and addition of epoxy-bonded concrete prisms characteristics [15]. An investigation results of mortars resistance to attack of magnesium sulphate revealed a notable alteration in their compressive and flexural characteristics [16]. investigated the effects of strong alkaline materials are bicarbonate and sodium carbonate in water mixtures of the concrete mix and characteristics of concrete strength. Therefore, it is necessary to evaluate how different concentrations of magnesium chloride in water affect the cement's setting times and concrete's strengths [17].

Sodium chloride, sometimes known as rock salt, has a strong taste and is very soluble in water, making it a useful deicing agent for melting ice at temperatures below 20 °F [18]. When compared to other salts, it is comparatively safe for concrete [19]. Equal amounts of sodium as positively charged and chloride ions as negatively charged make up the ionic composition of sodium chloride [20]. Early research [21] on the effects of sodium chloride in mixing water is obtained an improvement of compressive strength at a concentration of 25 g/kg of solution and mild steel significantly decreased corrosion to transfer of water vapor. NaCl has been observed to have unpredictable effects on concrete, accelerating set in certain cement and delaying it in others [22].

Research on NaCl-containing deicers shows that NaCl significantly reduces concrete's compressive strength [23]. The concrete is affected by the corrosion in the zone of chloride salt solution obtained at a corrosion rate is 1.74 mm/year when sea water has a NaCl level of roughly 3.5% [24], 5% of NaCl solutions are utilized to research work the properties of durability at RHA-containing concrete in a laboratory setting [25]. The recognized qualities of concrete may be improved by adding Rice Husk Ash (RHA) [26-28]. At 28 days, the addition of RHA at 10% to reduced chloride ion penetration in concrete compared to cubes. A pozzolana called RHA is utilized all around the world to create high-performance concrete constructions. The primary component of increased concrete performance is dense calcium silicate hydrates (CSH), which are created when reactive amorphous silica in the concrete matrix combines with the release of cement hydration by calcium hydroxide [29, 30]. According to the results, Ca²⁺ and OH⁻ ions combine in the RHA of silica amorphous silica to generate less portlandite and more CSH gel, which increases the concrete strength of RHA and non-RHA-containing concrete [31].

For both financial and safety considerations, the primary issue with concrete constructions is their long-term endurance. Aggressive chemicals seeping into the concrete is typically the cause of concrete structure degradation [32]. Many concrete constructions, including power plants, waste disposal facilities, decks, piers, floating offshore platforms, and harbors, are frequently exposed to hostile environments. Carbon (IV) oxide, sulphates, chlorides, and moisture are a few examples of these aggressive agents [33]. The most serious issue impacting the longevity of concrete structures, particularly in salty settings, the reinforcing steel bars is corrosion of reinforced concrete (RC) due to the entry of chloride ions. Pitting corrosion will happen when there is growth of chloride on the implanted steel rebar surface [34]. Worldwide, RC corrosion is an issue that requires large sums of money for restoration and repair. Numerous studies have been conducted to determine the correlation among the concentration of chloride ions and the commencement of corrosion [35-39].

The concrete pore solution-free penetrates are not affected by all chloride ions, as shown by Haque and Kayyali [40]. When some of the ions are chemically bonded by the hydration products, chloroaluminate hydrate is produced. The RC structures are broken down by the portion of the chloride ions that remain free. According to Sague's [41], the control of depassivation in steel rebar concrete is used as the ions of chloride and hydroxide. However, they have recommended that while corrosion will happen at a reduced rate when internal chlorides are present, a ratio of less than 3 does not significantly contribute to corrosion when external chlorides are present [42]. Sulfurate assault is a term used to describe the detrimental effects of exposing materials of cement-based construction of an environment that contains sulphate ions. This is considered to be a major problem [43]. Researchers have investigated the construction-based aggressive ions. They have collected the sources from different zones such as deicing salts, ocean, groundwater, infectious activity, unwanted water from companies, usage of chemical contaminated mixed water [44–47].

3. Materials and Methods

3.1 Materials

OPC was utilized to this investigation. Materials for geopolymer concrete (GPC), including coarse aggregate, sand, fly ash, calcined clay, sodium hydroxide (SH), sodium silicate (SS), water, and super plasticizer (SP), are used in the current study project. Micro silica is also added to geopolymer concrete. For instance, A35 is a geopolymer concrete mixture in which 35% of the total binder is alkaline activator solution (A). Table 1 displays the several binder combinations for the SS and SH, ranging from 1.5 to 2.5.

Mixture	GPC –	GPC –	GPC –	GPC –	GPC – 5	GPC –	GPC - 7	GPC –	GPC –	ОРС
Label	A35	A35	A35	A35	A35	A35	A35	A35	A35	OPC
Labei	R2.5	R2.5	R2.5	R2	R2	R2	R1.5	R1.5	R1.5	OPC
Aggregate as Coarse	1222	1222	1216	1222	1222	1222	1222	1216	1216	1054
Sand	658	658	655	658	658	658	658	655	655	740
Fly Ash	400	-	-	400	-	-	400	-	-	-
GGBFS	-	400	-	-	400	-	-	400	-	-
Calcined Clay	-	-	400	-	-	400	-	-	400	-
Cement	-	-	-	-	-	-	-	-	-	376
SH	40	40	40	47	47	47	56	56	56	-
SS	100	100	100	94	94	94	84	84	84	-
Water	8	8	8	8	8	8	8	8	8	151

 Table 1: Binder Ratio of SS and SH

SP	6	6	6	6	6	6	6	6	6	4.2

Based on the results of Table 2's maximum compressive strength (48.9 MPa) and the above binder composition, we have determined the optimal binder composition (2.5), which is then used to determine the fraction of the geopolymer concrete mixture.

Table 2: Compressive strength of binder ratio

Mix ID	Label	Compressive S	trength (MPa)
MIX ID	Label	28 days	90days
GPC – 1 (F)	A35 R2.5	42.3	43.4
GPC – 2 (G)	A35 R2.5	44.8	48.9
GPC – 3 (CC)	A35 R2.5	39.8	40.8
GPC – 4 (F)	A35 R2	40.5	41.4
GPC – 5 (G)	A35 R2	43.1	43.9
GPC – 6(CC)	A35 R2	38.6	39.7
GPC – 7 (F)	A35 R1.5	39.2	40.1
GPC – 8 (G)	A35 R1.5	41.9	43.1
GPC – 9 (CC)	A35 R1.5	37.8	38.7
OPC	OPC	38.9	41.5

3.2 Mixture Proportion

The concrete mixes are in two categories: initially with fly ash, GGBFS, and micro silica added, and another is pure 100% OPC. GGBFS and fly ash for various geopolymer blends. Superplasticizer and water were added to the mixes to make them more workable. Table 3 displays the mixture proportions of the ten blends that were examined. Combining solutions of SH and SS served as an alkaline activator. About half an hour before the concrete was actually mixed, the SS and SH solutions were combined used as alkaline activator. First, the aggregates and fly ash were combined in a pan mixer. To create fresh geopolymer concrete, the activator solutions were added to dry ingredients and mixed for an additional three to five minutes. A consistent ratio of SS/SH was used in creation of mixtures GPC1–10. A 35% alkaline activator was employed in each batch. Standard Portland Cement Concrete was designed using ACI 211.1 as a guide.

Table 3: Mixture Proportion (Kg/m³) of the concrete

Mixture	GPC -1	GPC -2	GPC -3	GPC -4	GPC -5	GPC -6	GPC -7	GPC -8	GPC -9	GPC - 10	ОРС
Label	A35	OPC									
Label	R2.5	Orc									
Coarse Aggregate	1222	1222	1216	1222	1222	1222	1222	1216	1216	1216	1054
Sand	658	658	655	658	658	658	658	655	655	655	740
Fly Ash	40	80	200	-	-	-	40	40	-	-	-
GGBFS	360	320	200	360	320	200	320	340	320	340	-
Calcined Clay	-	-	-	40	80	200	-	-	40	40	-

Micro Silica	-	-	-	-	-	-	40	20	40	20	-
Cement	-	-	-	-	-	-	-	-	-	-	376
SH	40	40	40	40	40	40	40	40	40	40	-
SS	100	100	100	100	100	100	100	100	100	100	-
Water	8	8	8	8	8	8	8	8	8	8	151
SP	6	6	6	6	6	6	6	6	6	6	4.2

3.3 Casting and curing

100 mm-sized concrete cubes were made in order to measure compressive strength. A total of 297 specimens were created using various geopolymer concrete compositions. After a 24-hour casting period, concrete cubes were demolding, and samples were submerged in water for 28 days to achieve the intended strength. Following this, 90 GPC 1–10 specimens and 90 GPC 1–10 specimens were moved into each of the 10% Na₂SO₄ and 10% NaCl solutions. The residual samples were then left in water for subsequent hardening at 28, 56, and 90 days. Three distinct exposure conditions were applied to the specimens: 10% Na₂SO₄ solution, 10% NaCl solution, and water. Table 4 has a detailed list of the specimens.

C	Curing Time	Absorbed under				
Samples Details	(Days)	Water	10% NaCl			
GPC-1-10 (For each composition 3 sample each)	28	30	30	30		
	56	30	30	30		
	90	30	30	30		
	28	3	3	3		
OPC	56	3	3	3		
	90	3	3	3		
Sub Tota	.1	99	99	99		
Total number of S	Specimens		297			

Table 4: Details of Samples

3.4 Testing of concrete specimens

Cast geopolymer concrete specimens of diameter and height is 100 mm and 200 mm were allowed to cure naturally at temperature $15\text{-}20 \,^{\circ}\text{C}$ and $70\pm10\%$ of relative humidity. For testing of compressive strength and mass change, cylinder specimens with diameter and height are $100 \, \text{and} \, 200 \, \text{mm}$ respectively. The weight change testing, the prism samples were submerged in $10\% \, \text{NaCl}$ and $10\% \, \text{Na}_2 \text{SO}_4$. The cylinder samples were submerged in the same chemical ratio to determine changes in compressive strength and mass.

4. Results and Discussion

4.1 Mechanical Properties

The compressive strength of concrete results are with and without reinforcement of fly ash shown graphically in Figures 1-3. The performance of GPC-7 (53.8 MPa) under water curing was found to be higher than that of OPC (38.9 MPa) at the 28-day mark. However, at later ages, GPC-7's compressive strength is 2.5% and 5.76% higher, respectively, than OPC's at 56 and 90 days. It was noted that fly ash and micro silica were added after 56 days and that this process grew steadily as the concrete aged. In addition, Figure 2 indicates that the performance of concrete containing GPC and OPC was determined to be equal under the 10% sodium sulphate (Na₂SO₄) exposure condition at 28 days, 56 days, and 90 days respectively. For short-term exposure conditions, it was found that the sodium sulfate solution had no discernible effect on the concrete, either with or without fly

ash. Similar outcomes were seen by investigated OPC over a maximum 360-day exposure time in a Na_2SO_4 solution [48]. They have used fly and bottom ash, and furnace slag in place of some cement. They found that the mechanical property of blended cement was reached 2% higher than OPC following a 360-day exposure to Na_2SO_4 .

It was also reported that, even after 300 days, a reduced concentration of Na₂SO₄ did not significantly alter the qualities of the mortar. Furthermore, the mechanism underlying the Na₂SO₄ reaction was previously understood to be that sulfate ions penetrate within concrete pores, and that a chemical interaction might occur between sulfate ions and cement hydration. Na₂SO₄ then reacts with CaO to produce mono-sulfate, which in turn forms gypsum and ettringite within the pores of the concrete [49]. Even though ettringite development is not ideal over the long run, it can be inhibited by adding fly ash since it causes the pores in the concrete to shrink. Furthermore, the improvement of GPC strength was observed very slowly when exposed to 10% NaCl compared to OPC. Nonetheless, the pore sizes circulation is crucial at hardened concrete and is impacted by the occurrence of chloride ions. Decalcifications that are obvious at later ages cause degradation and the production of chloro-aluminate in chloride solutions. The effects of Na₂SO₄ on concrete include decalcification, the creation of porous CaSO₄, and calcium hydroxide leaching.

The leaching of calcium hydroxide (CaO) is a significant phenomenon in concrete; however, the influence of fly ash, which contains amorphous silica, reacts with Na₂SO₄ that is generated when cement hydrates, lowering the amount of CaSO₄ that is present in the concrete overall. Thus, the physical appearance of concrete is ultimately impacted by the presence of chlorides, which also alter pore diameters and disrupt the hydration process. Given the information regarding the impact of chloride on concrete, the experimental results showed that OPC in a 10% NaCl solution gradually increased strength at 56 days and decreased strength later 90 days of contact. Though it was discovered that geopolymer concrete performed better than OPC, it was a noticeable material strength has been increased with time. Therefore, it was previously recognized that while the sulphate attack product, ettringite, was difficult to generate, the consumed calcium oxide made the concrete denser. Fly ash, on the other hand, contains less calcium oxide, which may lessen the sulphate attack. Therefore, it was discovered through experimentation that fly ash-containing concrete has an appropriate compressive strength and remains unaffected by Na₂SO₄ solution. However, its chemical reaction slows down and takes longer to recover below 10% NaCl.

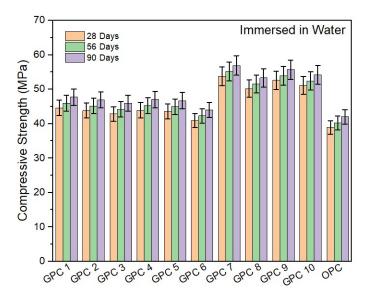


Figure 1. Schematic representation strength of OPC and GPC for immersed in water.

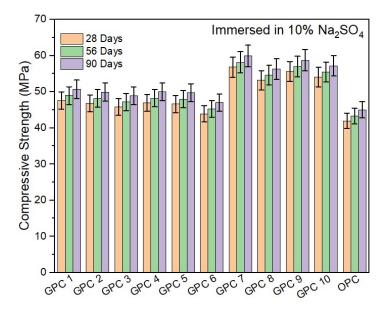


Figure 2. Schematic representation strength of OPC and GPC for immersed in Na₂SO₄.

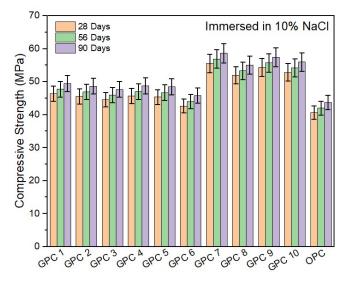


Figure 3. Schematic representation strength of OPC and GPC for immersed in NaCl.

4.2 Compressive Strength Variation

By comparing compressive strength of GPC to regular Portland concrete under various exposure of 28, 56, and 90 days respectively. The study is able to determine the strength of compressive variance in different sources. The concrete compressive strength containing fly ash was shown unchanged at 28 days, but to significantly increase at the ages of 56 and 90 days with solutions. Under Na₂SO₄ solution, fly ash-containing concrete performed compressive strength is better than others. After 90 days, it had reached a strength that was nearly 33.11% more than the other control mix exposure circumstances. Then, under a 10% NaCl solution, the concrete with fly ash depicted less compressive strength variance. Figure 4 depicts that the results of an examination of variation (%) in concrete compressive strength with fly ash and micro silica exposed to Na₂SO₄ and NaCl at varying curing days.

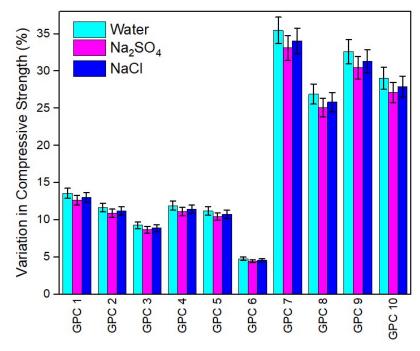


Figure 4. Variation of strength in OPC and GPC for immersed in water, sulphate and chloride solution.

4.3 Revolution in Weight

The concrete samples are weighing under water, Na₂SO₄, and NaCl was done both before and after exposure. Figure 5 shows the weight change of 10% fly ash-containing geopolymer concrete after 28, 56, and 90 days of curing. It was noted that after typical water curing, the weight of all types of concrete remained same. But there was a noticeable shift in weight in all combinations when it was exposed the solution at 10% Na₂SO₄ and NaCl. Under Na₂SO₄ at 56 and 90 days, the geopolymer concrete (GPC-7) showed the least weight gain; in the same conditions, other geopolymer concretes and OPC also showed greater weight gain.

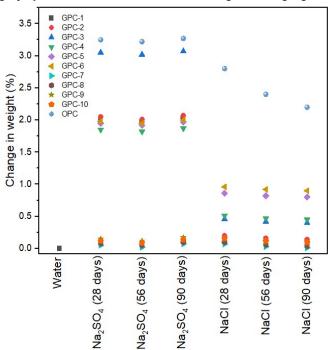


Figure 5. The concrete weight change under sulphate and chloride exposure.

It demonstrates how the penetrability of salts is decreased in concrete that contains fly ash and micro silica, with lower values observed compared to others. Less weight growth was seen in geopolymer concrete as a result of the findings that decreased fly ash and micro silica in the concrete might slow down process of hydration and lessen absorptivity of salts in concrete (GPC-7). It guaranteed that adding more cement to concrete would lessen the permeability of strong salt solutions, which corrode reinforcing and ultimately lead to structural collapse. Therefore, under 10% Na₂SO₄ and 10% NaCl circumstances, the overall performance of concrete decreased with fly ash and micro silica was shown satisfactory.

4.4 Degree of damage

According to the concept of damage degree, which is the indication of concrete deterioration, the damage degree was computed using the following equation, which Niu, [50], had previously determined.

$$D_i = 1 - \frac{\sigma_i}{\sigma_0} \tag{1}$$

On the other hand, D_i represents the extent of damage following a specific immersion time, σ_i represents the concrete's compressive strength following specific immersion duration, and σ_0 represents the concrete's original compressive strength. The mechanical properties value of OPC and GPC at 28 days of age, prior to being transferred into sulfate and chloride solutions, is represented by the σ_0 value in this study. The aforesaid Eq. (1) was used to assess the degree of damage for concrete with and without fly ash exposed to Na₂SO₄ and NaCl. The results are shown graphically in Figure 6. The findings showed that OPC experienced damage to a greater extent in all circumstances. In addition, compared to OPC and other geopolymer concrete, geopolymer concrete incorporating fly ash with micro silica (GPC-7) has reduced degree of damage. Additionally, Ming et al. [51] concurred that as damage increases, the capability of construction decreased. Eventually, when crack reaches are particular threshold, the structure concrete completely collapse. On the other hand, OPC -0.52, -0.57, and -0.54 in water, Na₂SO₄, and NaCl, respectively, showed greater results. Lower values were then found in geopolymer concrete under water, Na₂SO₄, and NaCl, respectively, measuring -0.57, -0.62, and -0.59. Therefore, it is stated here that the stronger and more durable the concrete, the lesser the degree of damage.

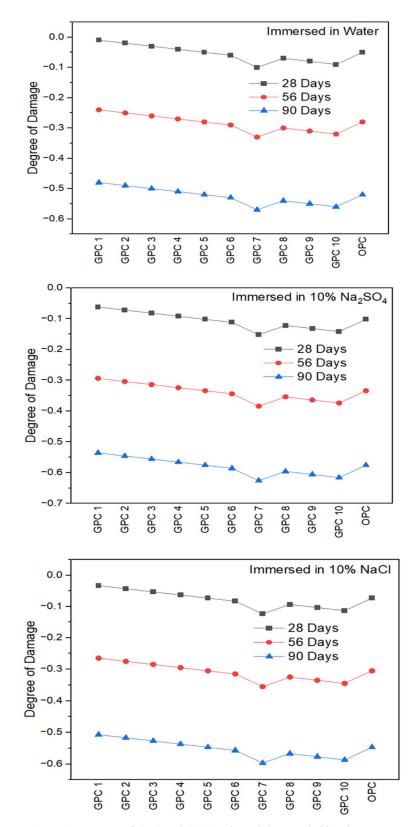


Figure 6. Damage of OPC and GPC under sulphate and chloride exposure at hardening time.

4.5 Ultrasonic Pulse Velocity (IPV)

Figure 7 depicts that the UPV of geopolymer concrete and regular Portland concrete varies with different days of curing for the concentration of alkali solutions. For one sample, UPV as the curing period increased (GPC-7). Nonetheless, in the samples at varying stages of maturation, there are differences in the velocity enhancement's magnitude and rate of growth. The sample containing GPC-7 reached the greatest velocity, followed by GPC-9 and GPC-8, in that order. The strength of GPC prepared with 10% fly ash is superior to that of other alkali concentrations, as evidenced by the rise in velocity for the latter. This study selected 90-day-old hardened concrete as the reference topic for examination in order to streamline the analysis process.

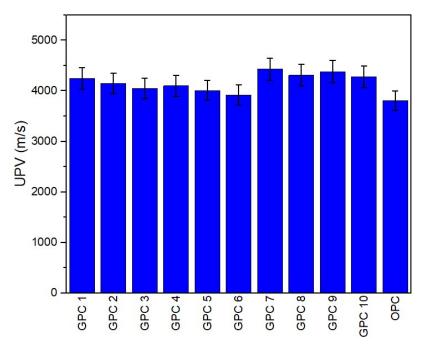


Figure 7. Progress of UPV at 90 days curing of GPC synthesized by different concentration of alkali activator. For geopolymer concrete, 90-day velocity trend was high. For GPC-7 and OPC, the maximum and lowest

ror geopolymer concrete, 90-day velocity trend was high. For GPC-/ and GPC, the maximum and lowest velocities have been noted, respectively. When accounting for the minimum and greatest parameters of standard deviation, which are comparable to mechanical properties that has been predictable at various curing periods, it is nearly constant after 90 days of maturity. The experimental results show that a higher alkali content led to a stronger initial polymerization. When concrete ages and the concentration of alkali activator is high, the discharges of additional humidity from the framework of geopolymer precursor. The structure becomes porous and fragile due to this moisture loss, which slows down the polycondensation reaction portion.

4.6 Rapid chloride permeability test (RCPT)

During experiment, the conductance of electrical on GPC was recorded for approximately six hours through charge passing. The amount of charge passed through test samples was calculated using Eq. (2), which was derived using the trapezoidal law and data collected every 30 minutes.

$$Q = 900 (I_0 + 2I_{30} + 2I_{60} + \dots + 2I_{330} + I_{360})$$
 (2)

Where, Q = Charge passed (Coulombs)

 $I_0,\,I_{30},\,I_{60}...I_{330},\,I_{360}$ – Current at 0, 30, 60...330, 360 minutes

According to ASTM C 1202 [53], the outcome of chloride ion dispersion relies on the charge transmitted. The chloride ion permeability of OPC and GPC mixtures is depicted in Table 5.

Samples	Composition	Average Current (coulombs)
GPC-1	F10+G90	1230
GPC-2	F20+G80	1310
GPC-3	F50+G50	1790
GPC-4	G90+C10	1480
GPC-5	G80+C20	1860
GPC-6	G50+C50	2230
GPC-7	F10+G80+S10	1045
GPC-8	F10+G85+S5	1160
GPC-9	G80+C10+S10	1150
GPC-10	G85+C10+S5	1215
OPC	OPC	2380

Table 5: RCPT outcome of OPC and GPC

It shows that the best way to resist Cl⁻ permeability in fly ash-GGBFS and micro silica-based GPC was increasing the percentage of composition. The charge applied to other geopolymer concrete and OPC is also decreased. The minimal permeability of the GPC-7 sample can be primarily ascribed to an increased incorporation of GPC. On the other hand, the concrete mix (GPC-6) is classified as high chloride ion permeability. Nevertheless, compared to all other mixes, the mix GPC-7 which consists of 80% GGBFS, 10% fly ash, and 10% S showed reduced chloride ion permeability. Additionally, the same composition has demonstrated increased compressive strength. This was credited to a skilful modification of the pore structure of geopolymer concrete.

4.7 Scanning Electron Microscopy (SEM)

Figure 7 depicts that the pictures from study of SEM with fly ash-containing sample that was exposed to $10\% \, \text{Na}_2 \text{SO}_4$ and $10\% \, \text{Na}_2 \text{Cl}$ over at the days of 56.

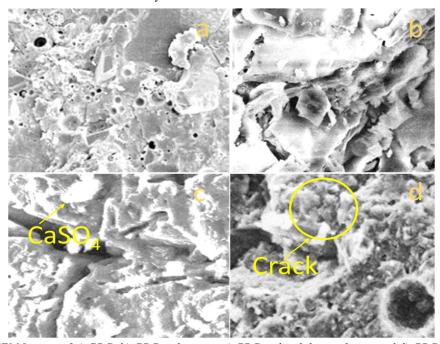


Figure 7. SEM Images of a) GPC; b) GPC with water; c) GPC with sulphate solution and d) GPC with chloride solution.

It is evident that the calcium sulphate reaction starts at 56 days of age, and a well-formed reaction between calcium oxide was seen. The concrete that contained fly ash also showed signs of calcium sulphate production. A small quantity of ettringite was fashioned in concrete while it was exposed to a solution of Na₂SO₄, which capacity partially fills in the voids and promotes the improvement of strength [52]. Similarly, the synthesis of calcium sulphate was observed in GPC when exposed to a Na₂SO₄ solution. But in this case, a notable illustration of a reduction in compressive strength was seen as a symptom of crack propagation.

5. Conclusions

From the outcome of the experimental results for OPC and GPC assumptions are the following;

- i. The outcome of this research work suggested that fly ash and micro-silica might substitute some of the OPC in both benign and harsh environments.
- ii. It was noted that during typical water curing, fly ash-infused concrete does alter in weight. However, a noticeable shift in weight was seen in all combinations are subjected to solution of 10% Na₂SO₄ and NaCl. In OPC, the weight gain was increased. The minimum gain of weight was seen in geopolymer concrete, though, because fly ash may slow down the process of hydration and lessen salts' capacity to material penetrate.
- iii. GPC-7 has a stronger performance than OPC, which was cured in water at the days of 28. GPC has a compressive strength that is 2.5% and 5.76% developed than OPC at 56 and 90 days, respectively.
- iv. It was discovered that, up to ninety days, the presentation of fly ash-containing concrete, GPC, and OPC exposed to 10% sodium sulphate (Na₂SO₄) was equivalent. For brief exposure, the Na₂SO₄ solution has no discernible effect on concrete.
- v. Ordinary Portland concrete gradually gains strength at 56 days and gradually loses strength over 90-day exposure period when placed in a 10% sodium chloride (NaCl) solution. Nonetheless, it was discovered that geopolymer concrete performed better than OPC, but there was a noticeable increase in strength over time (GPC-7).
- vi. The strength performance of the concrete is not negatively impacted by the partial replacement of cement with fly ash mixed with micro silica. Nonetheless, it performs well enough in typical water conditions and when exposed to sulphate and chloride.
- vii. There are technical and environmental advantages to using fly ash and micro silica in place of regular cement, which are crucial in the current context of sustainable development. Therefore, it is suggested that fly ash coupled with micro silica be used as a concrete material. It is also recommended that more research be conducted to ascertain the material's long-term strength and durability in the presence of seawater and the combined effects of sulphate and chloride.

References

- 1. Oyenuga, V., 'Reinforced Concrete Design'. ASROS Ltd, 2005, pp 4-6.
- 2. Wikipedia, 'Salt water and concrete', 2005.
- 3. Taylor, H.F.W 'Chemistry of Cements' Vol. 1 Academic Press UK and USA, 1964.
- 4. A.M. HELMENSTINE, (2005); Melting Snow and Ice with Salt. Chemistry.bout.com/cs/howthingswork/al/aa/2070a.html
- 5. S. ROBERTSON (2002); Concrete Hard Facts of Durable Structures. www.scienceinafrica.co.za/2002/sept/concrete.htm.
- 6. T.M CARTER (2005); Facts about Deicing Salts and Magnesium Chloride, www.askthebuilder.com/B251
- 7. C. LEE, J.F. BONACCI, and M.D.A. THOMAS (2000); Accelerated Corrosion and Repairs of Reinforced Concrete Column using Carbon Fibre, Canadian Journal of Civil Engineering. 125(3) 1-5
- 8. Shetty, M.S., 'Concrete Technology: Theory and practice'. S. Chand & Company Ltd, 2000, pp 28, 67, 119 and 230.
- 9. Falade F. 'International Conference on Structural Engineering Analysis and Modeling'. SEAM3. Kwame Nkruma University of Science and Technology, Ghana, 1997.
- 10. Valore R.C 'Insulating Concrete', ACI Journal, Vol 5, Nov. 1956.

- 11. Cholker, A.K. and Tantray, M.A., 2020. Strain-sensing characteristics of self-consolidating concrete with micro-carbon fibre. Australian Journal of Civil Engineering, 18(1), pp.46-55.
- 12. S.O. Ojoawo and A.M Oladejo (2013) An investigation into the effects of contaminated with chloride salts on the compressive strength of concrete. Proceedings of 2nd International Conference on Engineering and Technology Research, FET Conference 2013, LAUTECH, Ogbomoso, Nigeria (March 2013), 2: 1-10. ISBN: 978-2902-58-6.
- 13. Kejin W, Daniel EN and Wilfrid AN (2005) Damaging effects of deicing chemicals on concrete materials. Cement & Concrete Composites. 28(2), 173-188.
- 14. Gorninsi JP, Dal Molin DC and Kazmierczak CS (2007), Strength degradation of polymer concrete in acidic environments. Cement & Concrete Composites. 29(2), 637-645.
- 15. Adnan C, Turgay C and Ahmet EB (2009) Effects of environmental factors on the adhesion and durability characteristics of epoxy bonded concrete prisms. Construction & Building Materials. 23.2, 758.
- 16. Fikret T, Fevziye A, Sema K and Nabi Yuzer (1997) Effects of magnesium sulfate concentration on the sulfate resistance of mortars with and without silica fume. Cement & Concrete Res. 27(2), 205-214.
- 17. Venkateswara Reddy V, Sudarshan Rao H and Jayaveera KN (2006) Influence of strong alkaline substances (sodium carbonate and sodium bicarbonate) in mixing water on strength and setting properties of concrete. Indian J. Engg. & Material Sci.13(2), 123-128.
- 18. http://www.encyclopedia.com/topic/sodium chloride.aspx -accessed 17 June 2010.
- Cody, R. D., Cody, A. M., Spry, P. G., & Gan, G. (1996). Concrete Deterioration by Deicing Salts: An Experimental Study. Department of Geological and Atmospheric Sciences, 253 Science I, Iowa State University, Semisequicentennial Transportation Conference Proceedings. May 1996, Iowa State University, Ames, Iowa. -accessed 11 June 2010.
- 20. http://www.peterschemical.com/sodium-chloride/ -accessed 17 June 2010.
- 21. Cholker, A.K. and Tantray, M.A., 2021. Electrical resistance-based health monitoring of structural smart concrete. Materials Today: Proceedings, 43, pp.3774-3779.
- 22. Mattus, C.H., & Gilliam, T.M. (1994). A literature review of mixed waste components: sensitivities and effects upon solidification stabilization in cement-based matrices. A publication of the U.S department of Energy office of technological development. Washington D.C. 20585. -accessed 18 June 2010.
- Shi, X., Fay, L., Peterson, M.M., Berry, M., & Mooney, M. (2011). A FESEM/EDX investigation into how continuous deicer exposure affects the chemistry of Portland cement concrete. Construction and building materials, 25 (2), 957-966.
- 24. http://www.corrosion-doctors.org/Corrosion-by-Water/Types-of-water.htm -accessed 11 June 2010.
- 25. Anwar, M., Miyagawa, T., & Gaweesh, M. (2000). Using rice husk ash as a cement replacement material in concrete. WASCON 2000. Waste materials in construction. Proceedings of the fifth international conference. -accessed 17 June 2010.
- 26. Givi, A. N., Rashid, S.A., Aziz, F. N. A., & Salleh, M. A. M. (2010). Contribution of rice husk ash to the properties of mortar and concrete: A review. Journal of American science. accessed 17 June 2010.
- 27. Kartini, K., Mahmud, H.B., & Hamidah, M.S. (2010). Absorption and permeability performance of Selangor rice husk ash blended grade 30 concrete. Journal of engineering science and technology. School of engineering, Taylor's university college, 5 (1). -accessed 18 June 2010.
- 28. Reddy, D.V., & Alvarez, M. (2006). Marine durability characteristics of rice hush ash modified reinforced concrete. Fourth LACCEI international Latin American and Caribbean conference for engineering and technology (LACCET'2006). Breaking frontiers and barriers in engineering: education, research and practice. 21-23 June 2006, Mayaguez, Puerto Rico. -accessed 18 June 2010.
- 29. Giannotti de Silva, F., Liborio, J. B. L., & Helene P. (2008). Improvement of physical and chemical properties of Brazilian silica rice husk. Revista ingenieriade construction, 23 (1), -accessed 18 June 2010.
- 30. Rodriguez de Sensale, G., (2006). Strength development of concrete with rice-husk ash. Cement and concrete composites, 28 (2), 158-160.
- 31. Qijun, Y., Sawayama, K., Sugita, S., Shoya, M., & Isojima, Y. (1999). The reaction between rice husk ash and Ca (OH)2 solution and the nature of its product. Cement and concrete research, 29 (1), 37-43.
- 32. J. M. Marangu, J. K. Thiong'o, and J. M. Wachira, "Chloride ingress in chemically activated calcined clay-based cement," Journal of Chemistry, vol. 2018, Article ID 1595230, 8 pages, 2018.

- 33. P. Grassl, H. S. Wong, and N. R. Buenfeld, "Influence of aggregate size and volume fraction on shrinkage induced micro-cracking of concrete and mortar," Cement and Concrete Research, vol. 40, no. 1, pp. 85–93, 2010.
- 34. C. Alonso, C. Andrade, M. Castellote, and P. Castro, "Chloride threshold values to depassivate reinforcing bars embedded in a standardized OPC mortar," Cement and Concrete Research, vol. 30, no. 7, pp. 1047–1055, 2000.
- 35. M. Verma; M. Nigam "Investigation on the Effect of Curing Time on the Mechanical Properties of Geopolymer Concrete" AIP Conference Proceedings, 2721(1), 020031, 2023. ISSN No: 1551-7616 DOI: 10.1063/5.0154116
- 36. N. Jayanthi; T. Ghosh; R. K. Meena; M. Verma "Length and width of low-light, concrete hairline crack detection and measurement using image processing method" Asian Journal of Civil Engineering, 25(3), 2705–2714, 2024. ISSN: 2522-011X https://doi.org/10.1007/s42107-023-00939-0
- 37. M. Nigam; M. Verma "Prediction of compressive strength of nano-silica concrete by using random forest algorithm" Asian Journal of Civil Engineering, 25(7), 5205 5213 2024. ISSN: 2522-011X. https://doi.org/10.1007/s42107-024-01107-8
- 38. R. Kumar; M. Verma; N. Dev "Analysis of PCE-based Superplasticiser for the Different Types of Cement using Marsh Cone Test" EVERGREEN Joint Journal of Novel Carbon Resource Sciences & Green Asia Strategy, 11, 2, 665-672, 2024.
- 39. M Gupta; K Upreti; S Yadav; M Verma; M. Mageswari; A Tiwari "Assessment of ML techniques and suitability to predict the compressive strength of high-performance concrete (HPC)" Asian Journal of Civil Engineering, In Press, 2024. ISSN: 2522-011X. https://doi.org/10.1007/s42107-024-01142-5
- 40. M. N. Haque and O. A. Kayyali, "Free and water-soluble chloride in concrete," Cement and Concrete Research, vol. 25, no. 3, pp. 531–542, 1995.
- 41. A. A. Sagues, "Chapter 3-corrosion of metals in concrete," in Galvanized Steel Reinforcement in Concrete, S. R. Yeomans, Ed., pp. 71–86, Elsevier Science, Amsterdam, Netherlands, 2004.
- 42. S. Mundra, M. Criado, S. A. Bernal, and J. L. Provis, "Chloride-induced corrosion of steel rebars in simulated pore solutions of alkali-activated concretes," Cement and Concrete Research, vol. 100, pp. 385–397, 2017.
- 43. K. K. Aligizaki, "Mechanisms of ion transport in cement-based materials during the application of corrosion mitigating electrical techniques," in Proceedings of the NACE-2012-1542, pp. 1–15, NACE International, Salt Lake City, UT, USA, March 2012.
- 44. R. R. Hussain and T. Ishida, "Enhanced electro-chemical corrosion model for reinforced concrete under severe coupled action of chloride and temperature," Construction and Building Materials, vol. 25, no. 3, pp. 1305–1315, 2011.
- 45. M. Maes, K. Van Tittel boom, and N. De Belie, the efficiency of self-healing cementitious materials by means of encapsulated polyurethane in chloride containing environments," Construction and Building Materials, vol. 71, pp. 528–537, 2014.
- 46. M. Nigam; M. Verma "Effect of nano-silica on the fresh and mechanical properties of conventional concrete" Forces in Mechanics, 10 (2023), 100165. ISSN: 2666-3597. DOI: 10.1016/j.finmec.2022.100165
- R. Kumar, N. Dev, S. Ram, M. Verma "Investigation of dry-wet cycles effect on the durability of modified rubberised concrete" Forces in Mechanics, 10 (2023), 100168, 2023. ISSN: 2666-3597. DOI: 10.1016/j.finmec.2023.100168
- 48. U. Sharma; N. Gupta; M. Verma "Prediction of the compressive strength of Flyash and GGBS incorporated geopolymer concrete using artificial neural network" Asian Journal of Civil Engineering, 24 (8), 2837–2850, 2023. ISSN: 2522-011X DOI 10.1007/s42107-023-00678-2
- U. Sharma; N. Gupta; M. Verma "Prediction of Compressive Strength of GGBFS and Flyash based Geopolymer Composite by Linear Regression, Lasso Regression and Ridge Regression" Asian Journal of Civil Engineering, 24 (8), 3399–3411, 2023. ISSN: 2522-011X DOI: 10.1007/s42107-023-00721-2
- 50. M. Verma "Prediction of compressive strength of geopolymer concrete using random forest machine and deep learning" Asian Journal of Civil Engineering, 24 (7), 2659-2668, 2023. ISSN: 2522-011X DOI: 10.1007/s42107-023-00670-w

- 51. M. Verma "Prediction of compressive strength of geopolymer concrete by using ANN and GPR" Asian Journal of Civil Engineering, 24 (8), 2815–2823, 2023. ISSN: 2522-011X DOI: 10.1007/s42107-023-00676-4
- 52. M. Verma; M. Nigam "Experimental Investigation on the Properties of Geopolymer Concrete after Replacement of River Sand with the M-Sand" AIP Conference Proceedings, 2721(1), 020029, 2023. ISSN No: 1551-7616 DOI: 10.1063/5.0154113
- 53. M. Verma; M. Nigam "Effect of FRP on the Strength of Geopolymer Concrete" AIP Conference Proceedings, 2721(1), 020030, 2023. ISSN No: 1551-7616 DOI: 10.1063/5.0154114